

Solution Suspension And Colloid

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A colloid is a homogeneous solution with intermediate particle size between a solution and a suspension. Colloid particles may be seen in a beam of light such as dust in air in a "shaft" of sunlight. Milk, fog, and jello are examples of colloids.

Mixture - Elmhurst University

What is a Solution? A solution is a homogeneous mixture of two or more substances. It is called a homogenous mixture, because the composition is uniform throughout the solution. The components of a solution are mainly of two types, solutes and the solvent. Solvent dissolves the solutes and form a uniform solution.

Difference Between Mixture and Solution | Compare the ...

Mercury Shaken in Oil . Mercury is a metallic element that is liquid at standard temperature and pressure. Because of its liquid properties, the element can be mixed with oil to produce a suspension. The mercury particles will disperse throughout the oil when the solution is shaken, but the particles will never dissolve.

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