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An insulated container of negligible mass holds 0.600 kg of water at 45.0°C . You put a 0.0500-kg ice cube at -15.0°C in the water (Fig. 20.23). (a)

Calculate the final temperature of the water once the ice has melted. (b) Calculate the change in entropy of the system.

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charge of $-2.205 \mu\text{C}$ is
located at (3.055 m,
4.501 in), and a charge
of $1.800 \mu\text{C}$ is located

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Summary of Chapter

20 • Equipotential surfaces are those on which the electric potential is constant.

• The electric field is perpendicular to the equipotential surfaces.

• Ideal conductors are equipotential surfaces.

• A capacitor is a device that stores electric charge. • Capacitance:

James S. Walker

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Charged Objects After
a comb is rubbed on a
wool sweater, it is able
to pick up small pieces
of paper. Why does the
comb ... Solutions

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QUESTIONS

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matters Discuss which
of the following should
be included in the cost
of equipment: (a)
installation charges, (b)
freight charges, (c)
cost of building
foundations, (d) new
parts needed to
replace those damaged
while...

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$D = d \cdot 20 = (s \cdot 2) \cdot 20 = s$
 $40 = (4.0 \text{ cm}) \cdot 40 = 25$
cm REASONING For convenience, we can assign due east to be

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the positive direction
and due west to be the
negative direction.

Since all the vectors
point along the same
east-west line, the
vectors can be added
just like the usual
algebraic addition of
positive and negative
scalars.

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